

AS – 568 High Temperature Gas Dynamics
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General species data for thermodynamic calculations

Species	Theta_r [K]	Theta_v [K]	Q_el	Heat of formation at 298K [kJ/mol]
CO	2.8	3120	$1 + 2\exp(-93500/T)$	-110.527
O ₂	2.1	2270	$3 + 2\exp(-11400/T)$	0
N ₂	2.9	3390	$1 + 2\exp(-99600/T)$	0
NO	2.44	2740	$2 + 2\exp(-174/T) + 2\exp(-63300/T)$	+90.291
O			$5 + 3\exp(-228/T) + \exp(-326/T) + 5\exp(-22850/T)$	+249.173
N			$4 + 6\exp(-27658.7/T) + 4\exp(-27671.7/T) + 6\exp(-41492.4/T)$	+472.683
N+			$9 + 5\exp(-22055/T) + \exp(-47070/T)$	+1873.0
O+			$4 + 10\exp(-38620/T) + 6\exp(-58270/T)$	+1560.3
NO+	2.86	3419	$2 + 2\exp(-174/T) + 2\exp(-75090/T)$	+988.2
electron			2	
CO ₂	0.56	1915 961 961 3383	$1 + \exp(-90000/T)$	-393.522
C(g)			$9 + 5\exp(-10200/T) + \exp(-21700/T)$	716.670
H				217.999
H ₂	87	6340	$G_0 + g_1 \exp(-132030/T)$	0
OH	27.18	5378	$4 + g_1 \exp(-47060/T)$	38.987
H ₂ O	13.37 20.87 40	5259 2297 5409	$1 + g_1 \exp()$	-241.826
Ar			$1 + 5\exp(-134130/T)$	0

Species	Theta dissociation [K]	Theta Ionisation [K]
O ₂	59500	142000
N ₂	113000	181000
NO	75500	108000
O		158000
N		169000